



The physics of sterilization

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The point of view of physics

Physics looks for general laws

Biology has to deal with specificities

« generic » vs « specific »

Generalities on sterilization processes

- Pressure
- Temperature
- Water vap. relative humidity
- Concentration of the sterilizing agent
- Inactivation of μ -organisms on all surfaces of medical devices
- Surface phenomena: condensation / adsorption

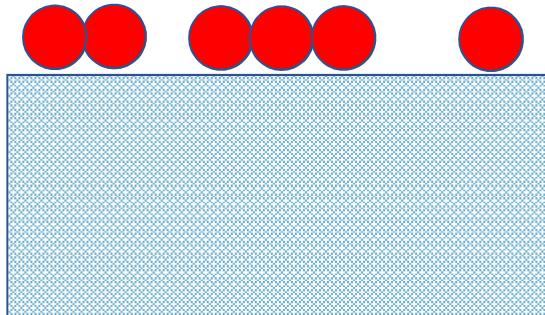
Agenda

1. Physics of adsorption and condensation
2. Review and discussion of the main sterilization processes (steam, ethylene oxide, formaldehyde, hydrogen peroxide)
3. A case study: H₂O₂ (microzoom,
*In partnership with microbalance and microbiological

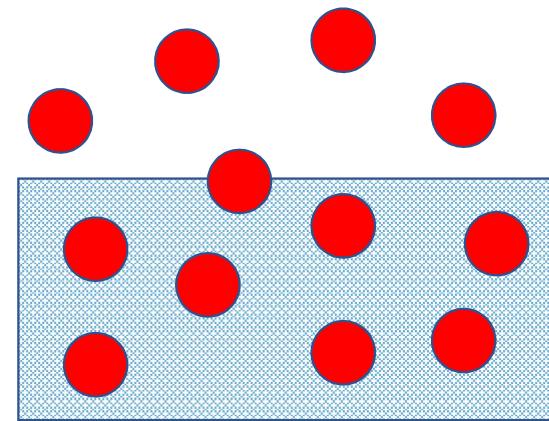
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1. Physics of adsorption and condensation

Adsorption, absorption, sorption



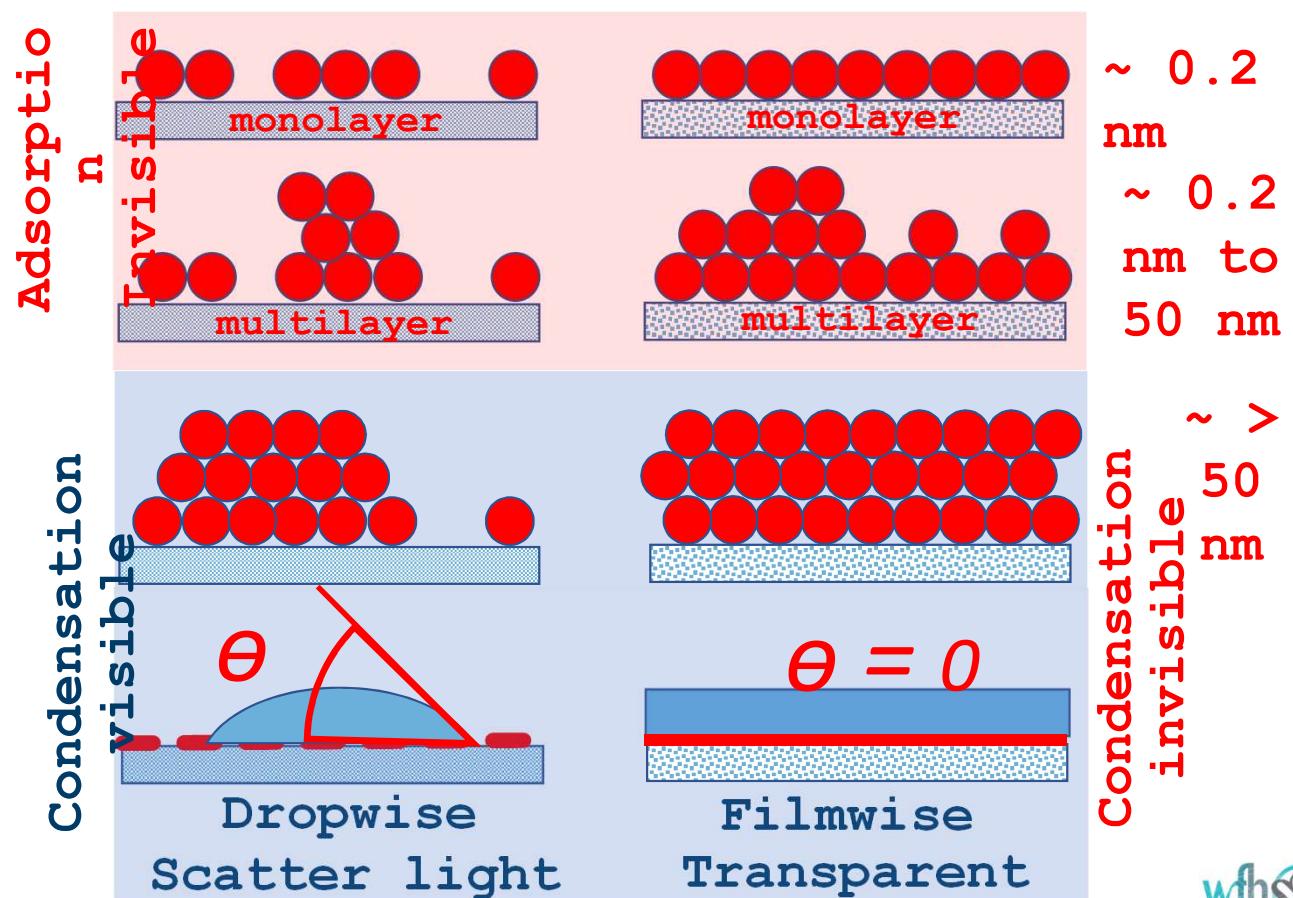
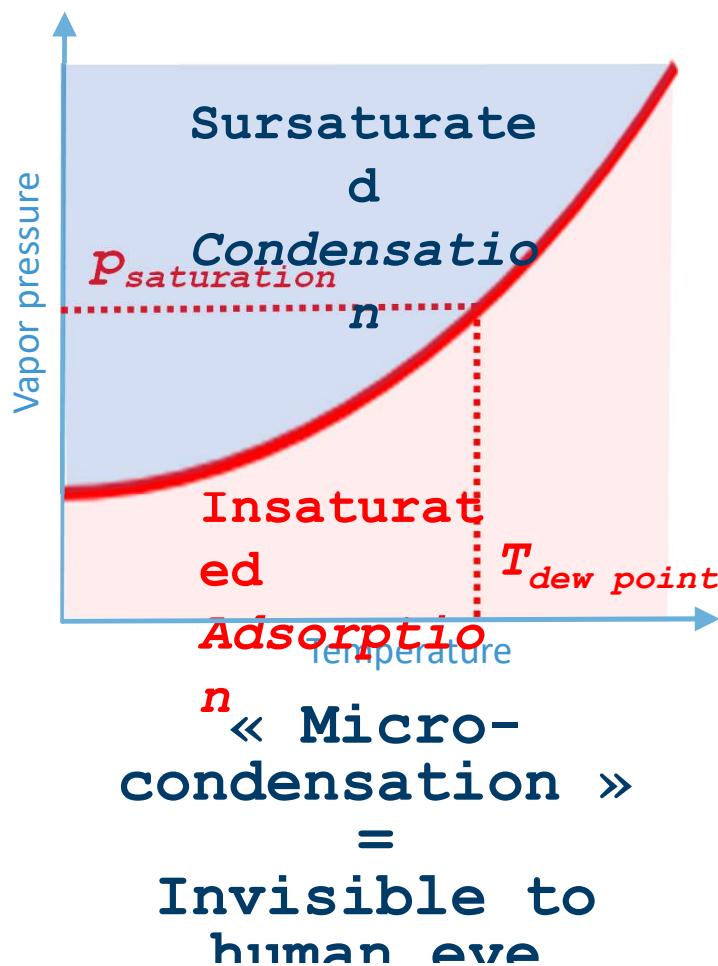
adsorption



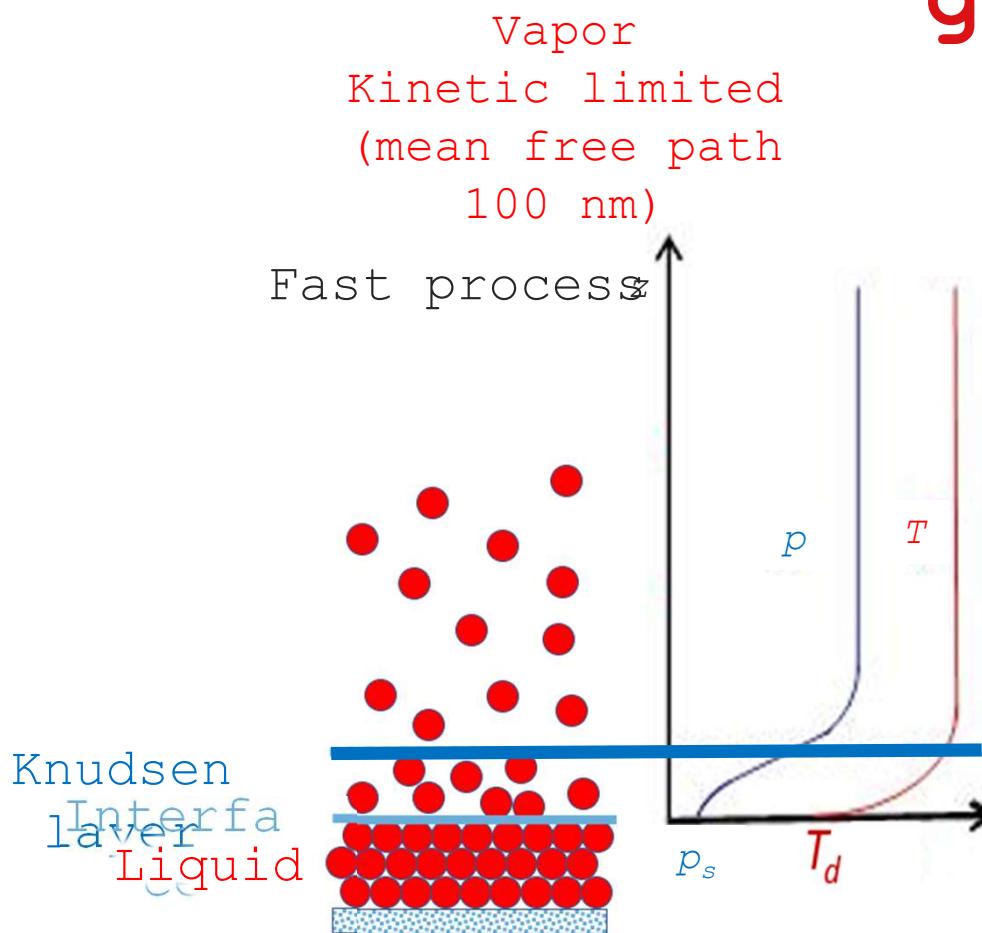
absorption

Sorption = adsorption + absorption

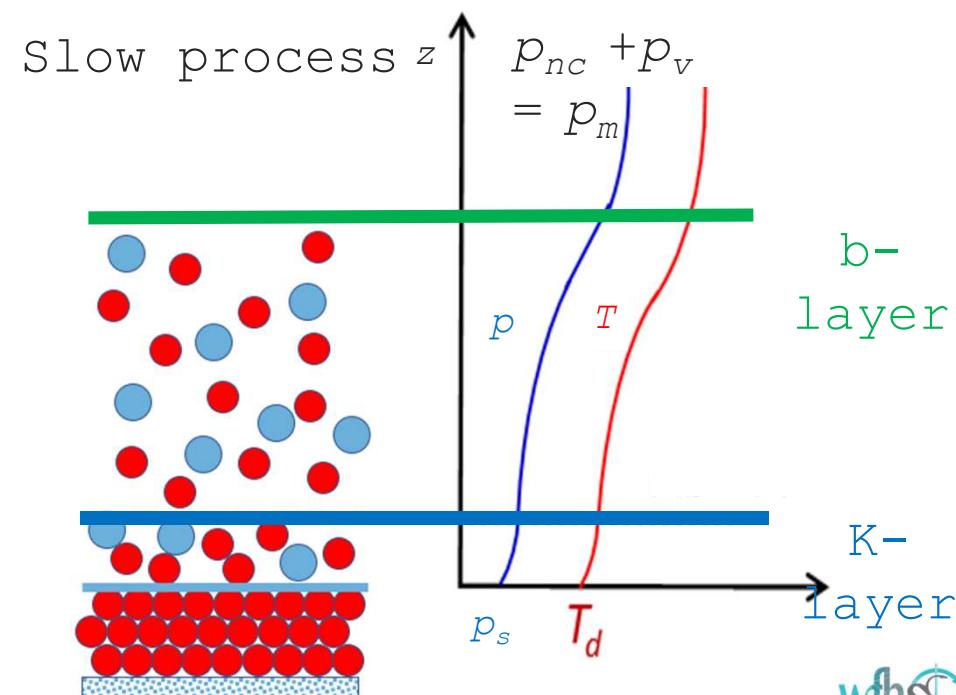
Adsorption/Condensation



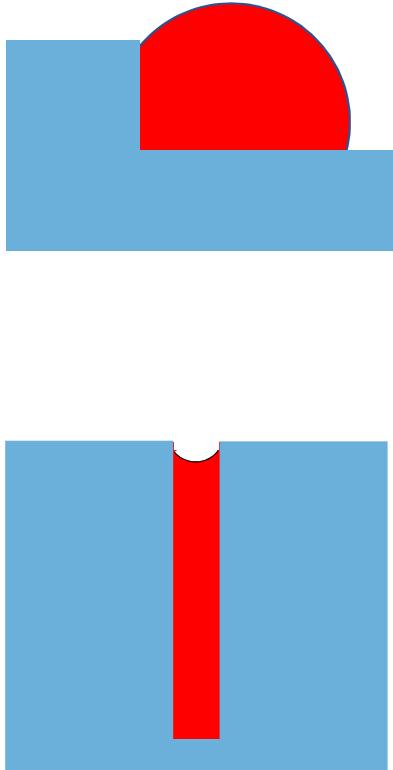
Why eliminate non condensable gases ?



Vapor + non-condensable gases
(e.g. air)
Limited by diffusions gradient
(boundary layer mm)



Role of surface geometry

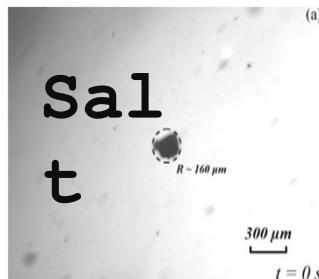


Enhanced nucleation

Surface energy is lowered

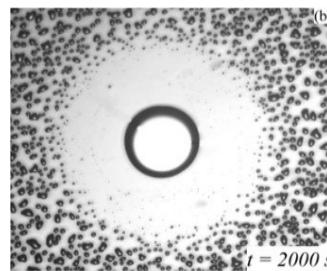
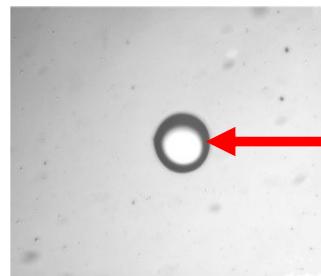
$p_c < p_s$; $T > T_d$

Hygroscopic materials



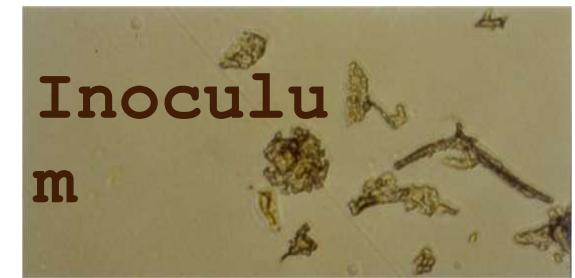
p_s (vap.+materials) <
 p_s (vapor)
 T_d (vap.+materials) >
 T_d (vapor)

Condensati
on

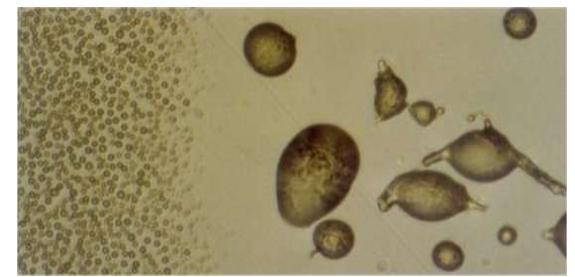


Salty water drop on
silanized glass; from
Guadarrama-Cetina et al.,
Pharm. Biopharm. 50 (2001) 18

The materials “pumps”
molecules around & dry
the surrounding,
preventing condensation



Lyophilized spores of *Bacillus*
macerans on silanized glass;
from Marcos-Martin et al.,
Pharm. Techn. Eur. 9 (1996) 1



Vapor mixtures

Adsorption

Condensation

- Different component concentrations (analogy with liquid solutions)
- Using Raoult's law and pure components isotherm
- Different component concentrations
- Using Raoult's law and pure components saturation pressures

2. Main sterilization processes

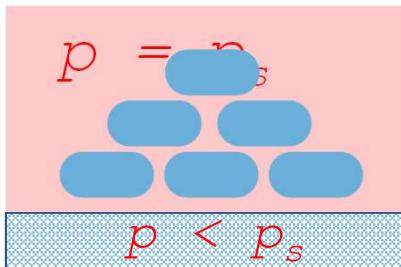
Generalities in

sterilization

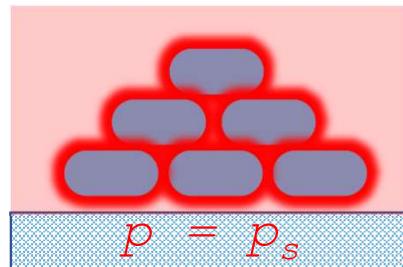
- Vacuum to eliminate non-condensable gas
- Gas phase to penetrate in load and instruments
- Condensation:
 - Allow penetration of sterilization agent in spores
 - Release of latent heat
 - Concentration of sterilizing agent in condensate phase
>> concentration in vapor phase ($\times 1000 - 10000$)
 - Liquid water + sterilizing agent: Modify dew point temp. to obtain high adsorption / condensation at sterilization chamber + ammonia (two phase diagram)

Saturated steam H₂O

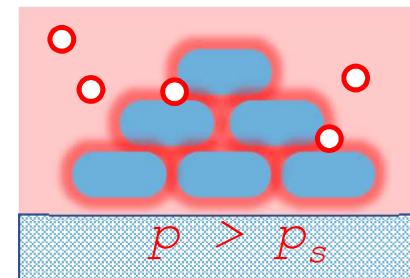
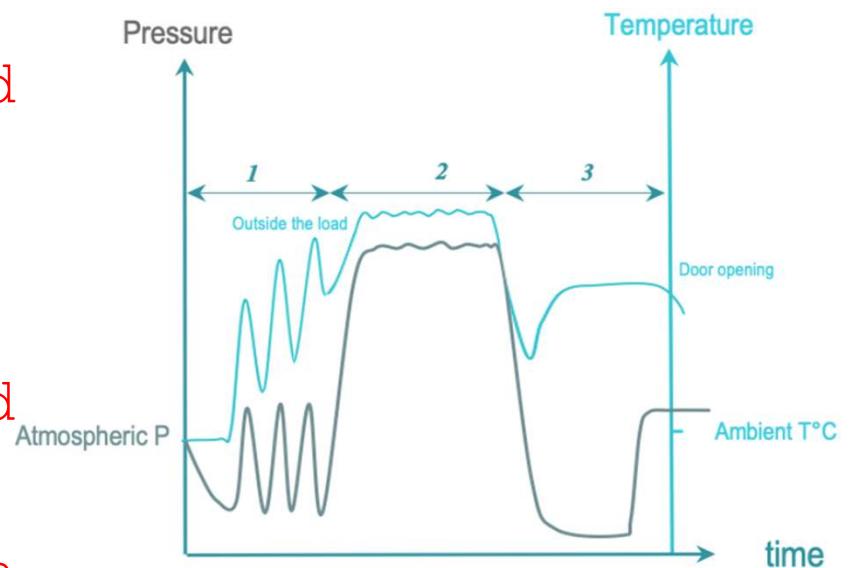
- Heat (121-134°C) + repeated latent heat releases by condensation
- Coagulation and denaturation of enzymes and structural proteins
- Steam exactly at saturation



No condensation



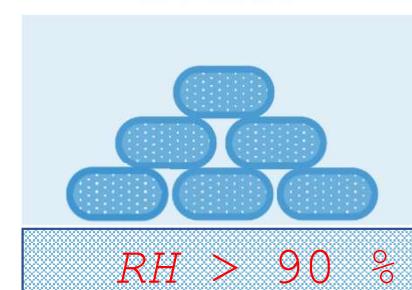
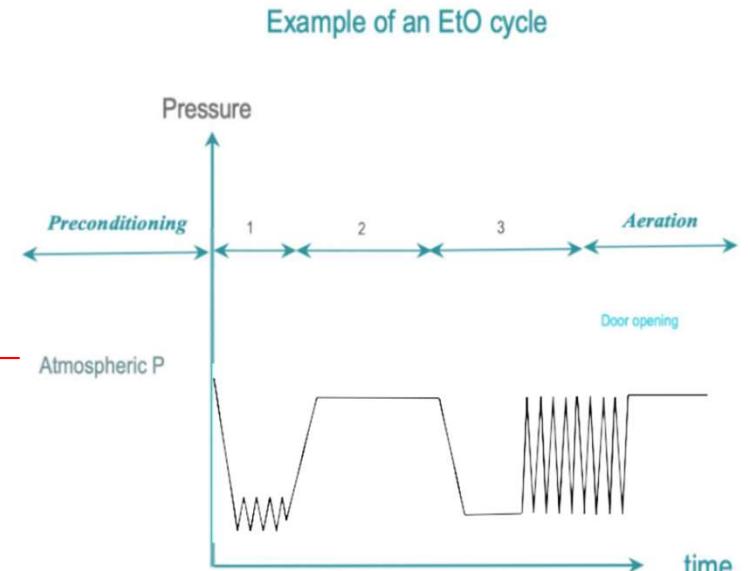
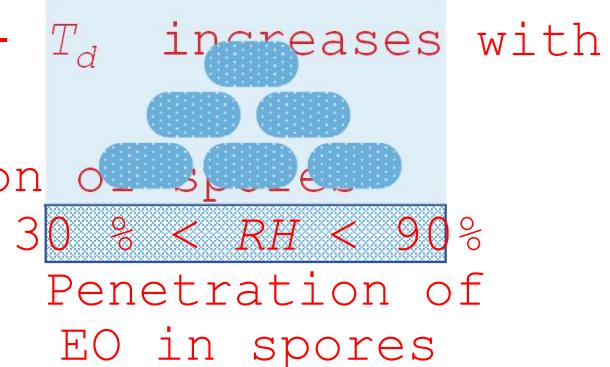
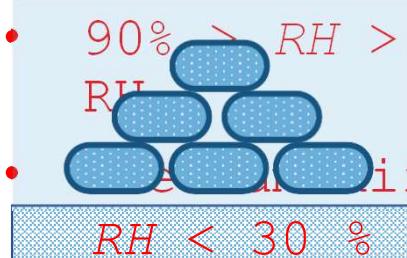
Condensation
Latent heat



% of vapor already
condensed

Ethylene Oxide C₂H₄O

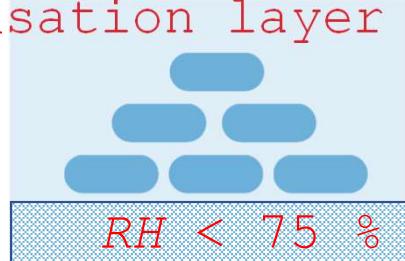
- EO: highly penetrative gas (absorption)
- Require aeration - carcinogenic
- Conc. 200-1200 mg/L
- Alkylation of protein, DNA and RNA - prion fixative
- Efficacy increases with T °C (typically 48 °C)



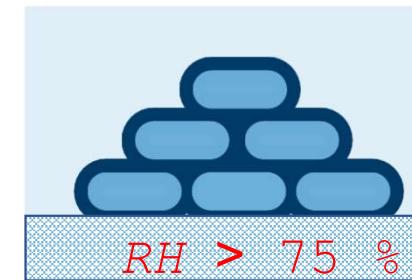
Efficacy vanishes

Steam formaldehyde CH₂O

- Temperature: 50 - 80° C
- Potentially carcinogenic – aeration
- Concentration: 15-100 mg/L
- RH >75%
- Alkylation of amino and sulfhydryal groups of proteins and ring nitrogen atoms of purine bases – prion fixative
- Condensation layer

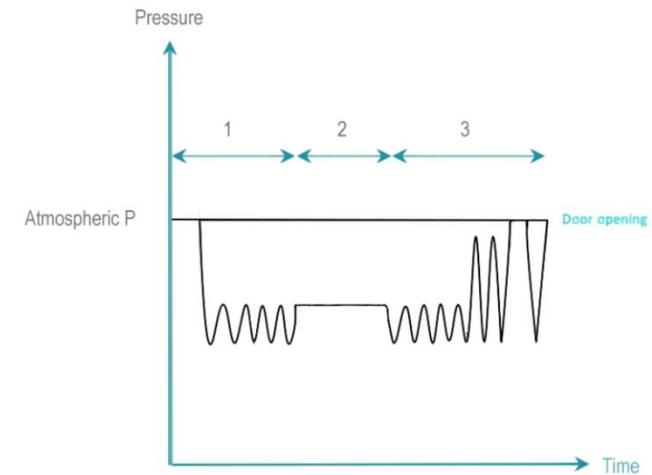


Not enough liquid around the spores



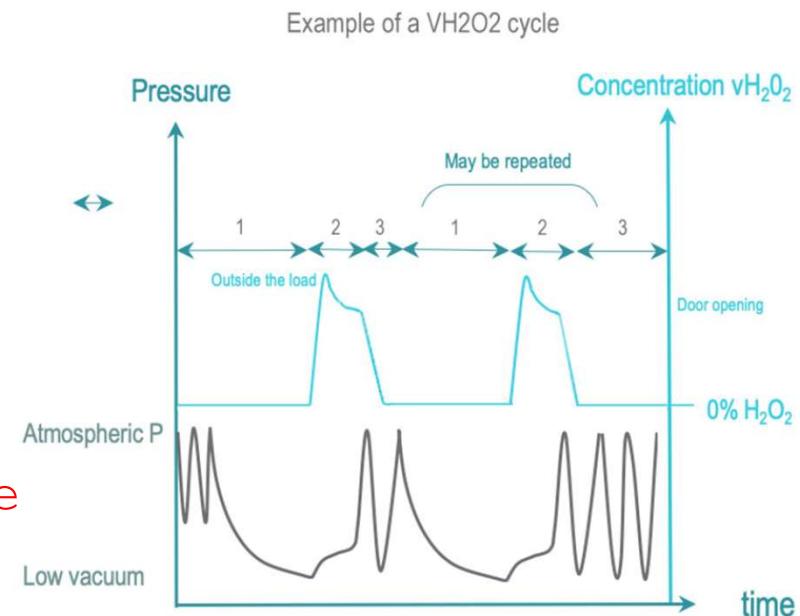
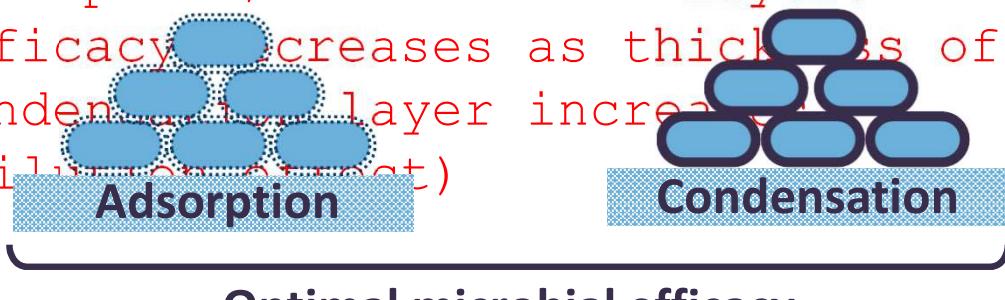
Condensation layer
Liquid phase (ISO 11138-5)

Example of an LTSF cycle

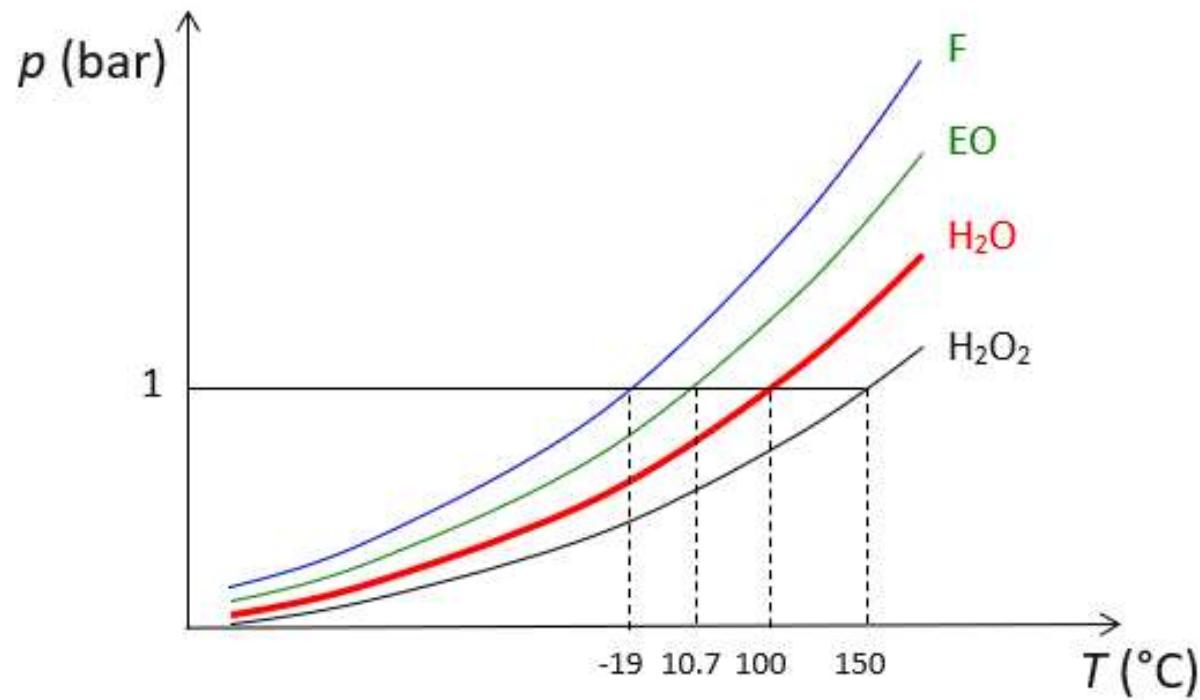


Hydrogen Peroxyde H₂O₂

- Injection of water + H₂O₂ after vacuum (\neq atmospheric pressure)
- Temperature < 60 °C
- Hydroxyl free radicals attacking membrane lipids, DNA, and other essential cell structures. Effective on prion.
- Adsorption/condensation layer:
Efficacy increases as thickness of condensation layer increases (dilution effect)



Sterilizing agents – T_{eb} .



3. Case study*: H₂O + H₂O₂

- Ambiant pressure and low cc.
H₂O₂: Slow motion
 - Influence of RH
 - Micro-zoom, micro-balance and
bio-controls
- * In partnership with
ASP (2012)

Test protocol

H₂O₂ sensors



Inactivation
kinetic

Pre-conditioning: RH 30%
to 80 %



Injection of µ-droplets
(evaporating)
H₂O, H₂O₂ at 6% and 12 %

Macro-zoom



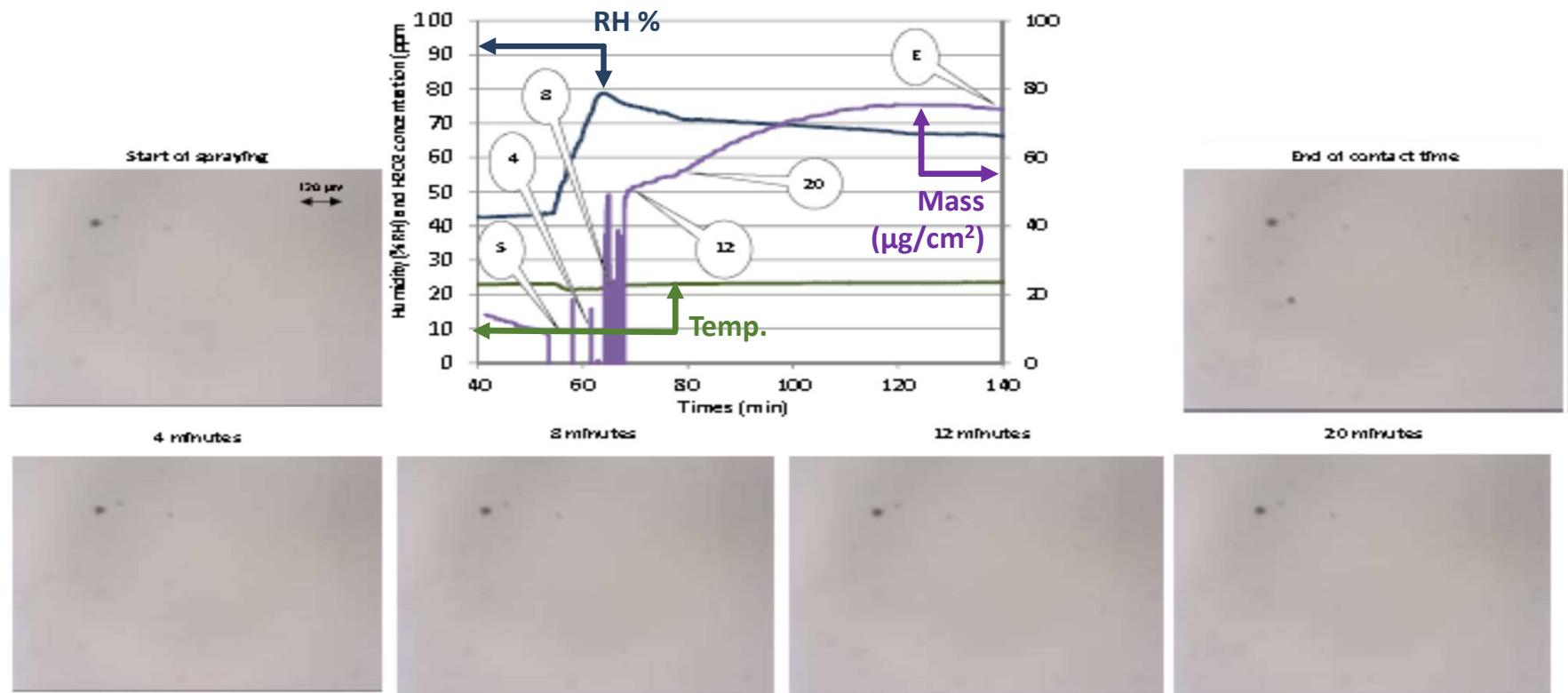
µ-
balance



WFHSS

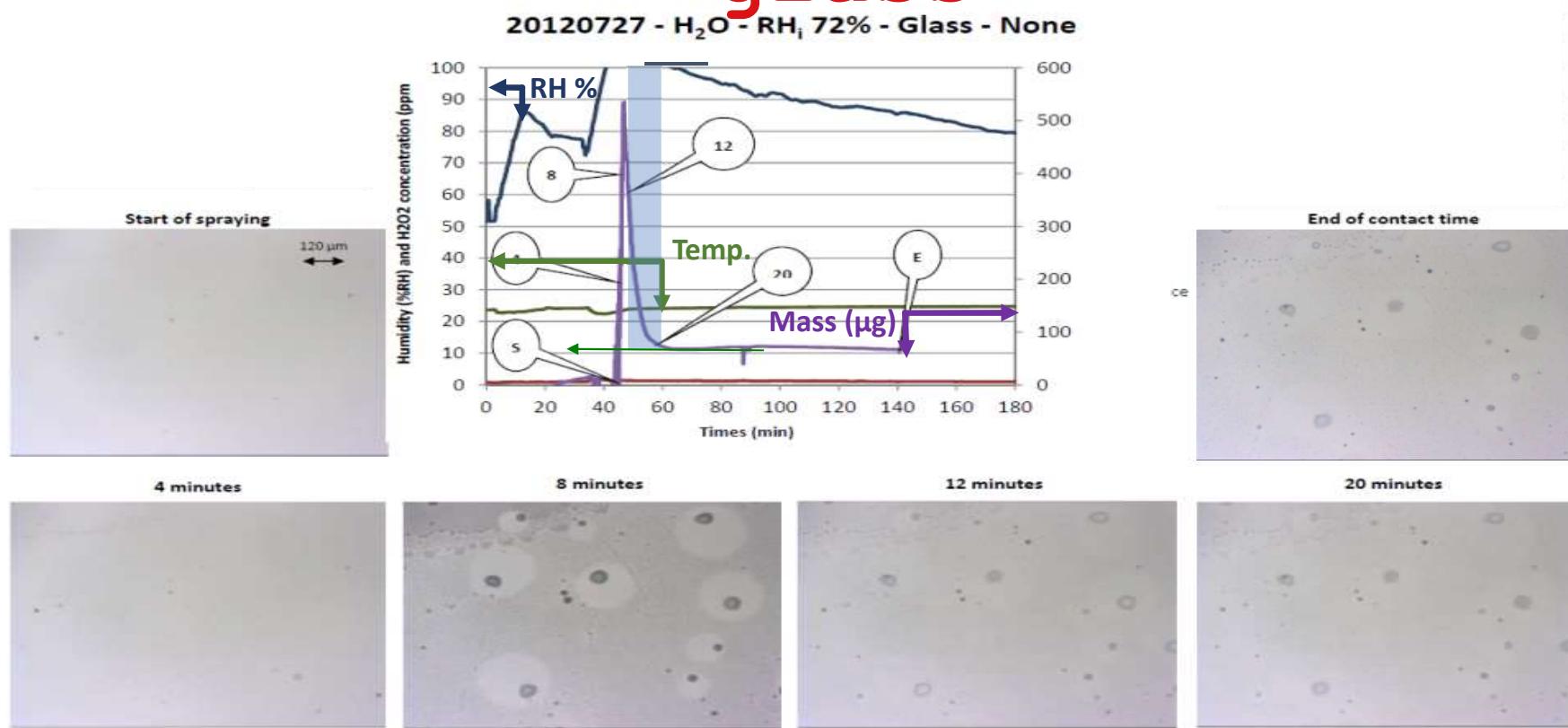
H₂O - Initial RH (41%) - glass

20120806 - H₂O - RH, 41% - Glass - None



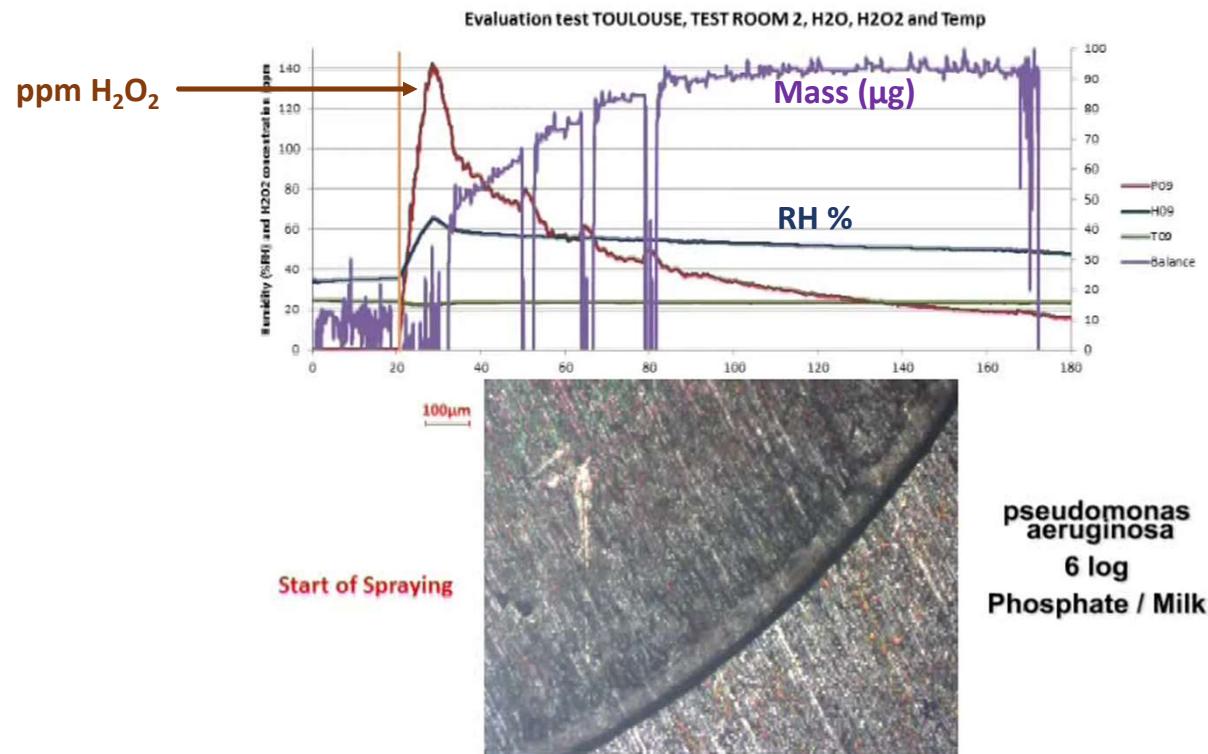
Adsorption and/or invisible
condensation

H₂O - Initial RH (72%) - glass

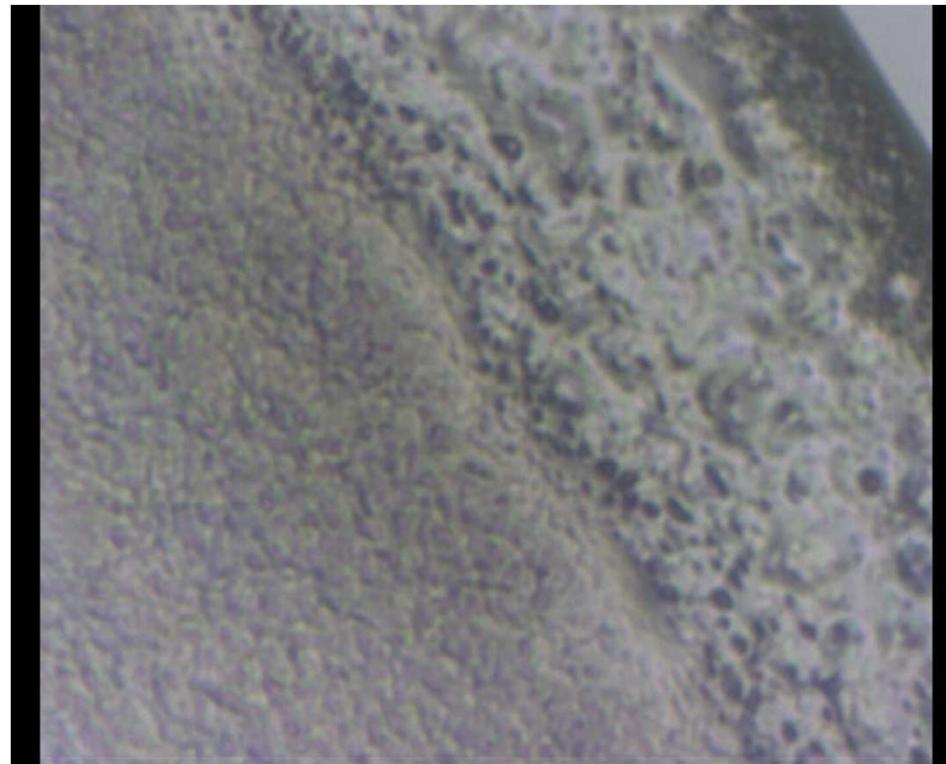


Visible dropwise condensation + rain

H₂O₂ 6 % - RH_i (35 %) - Germs on steel

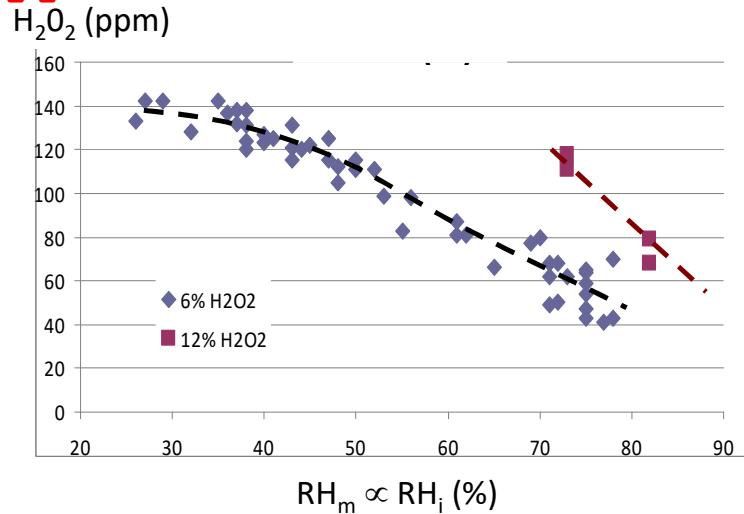


H₂O₂ 12 % - RH_i (51 %) -
Germs on steel

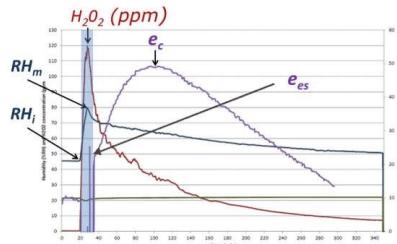
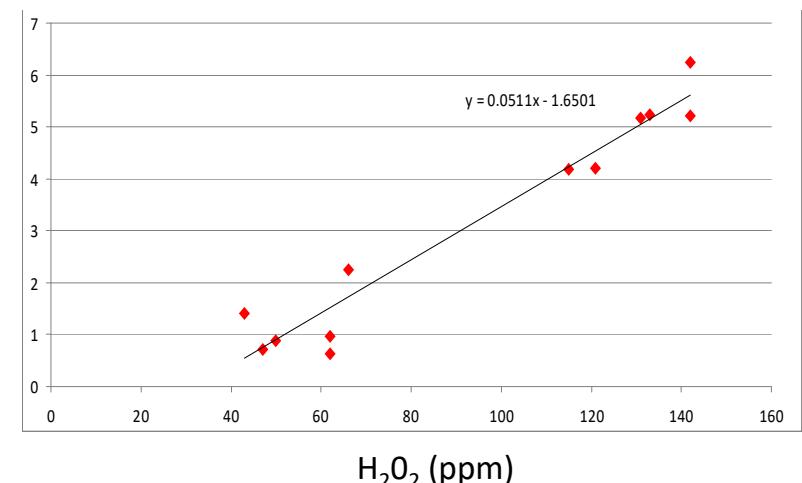


Case study - Analysis (1)

RH_m (RH_i) determines H_2O_2 ppm



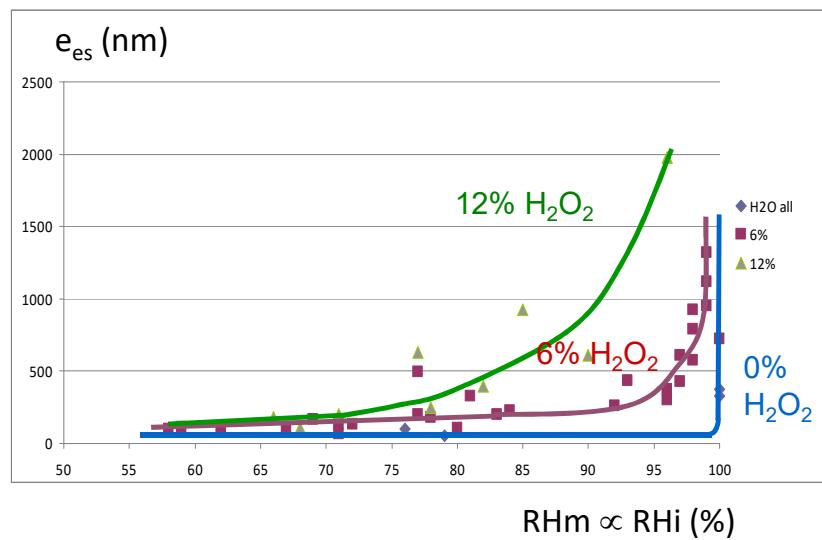
Efficacy ↑ when H_2O_2 ppm ↑ or RH_m (RH_i) ↓



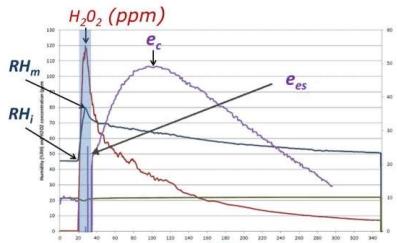
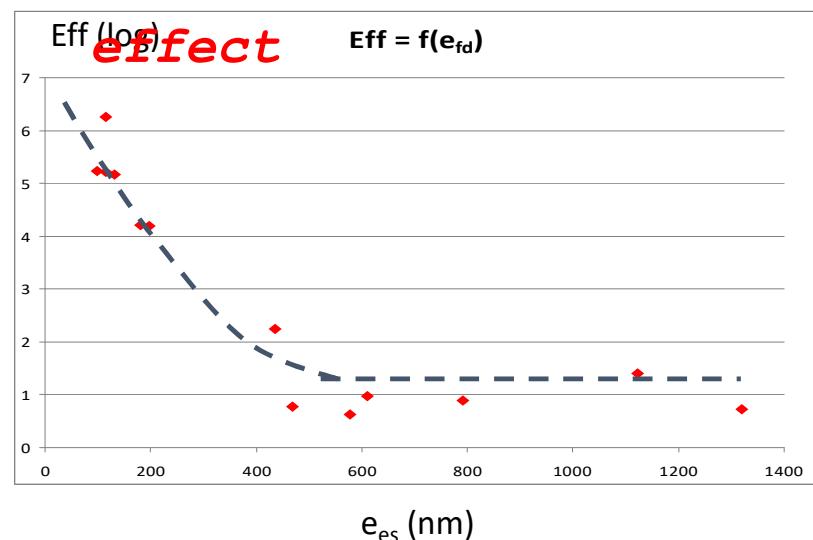
H_2O_2 liquid phase cc >> vapor phase cc
(Raoult and Henry's laws)

Case study - Analysis (2)

$e_{es} \uparrow$ $RH \uparrow$ or H_2O_2 ppm \uparrow



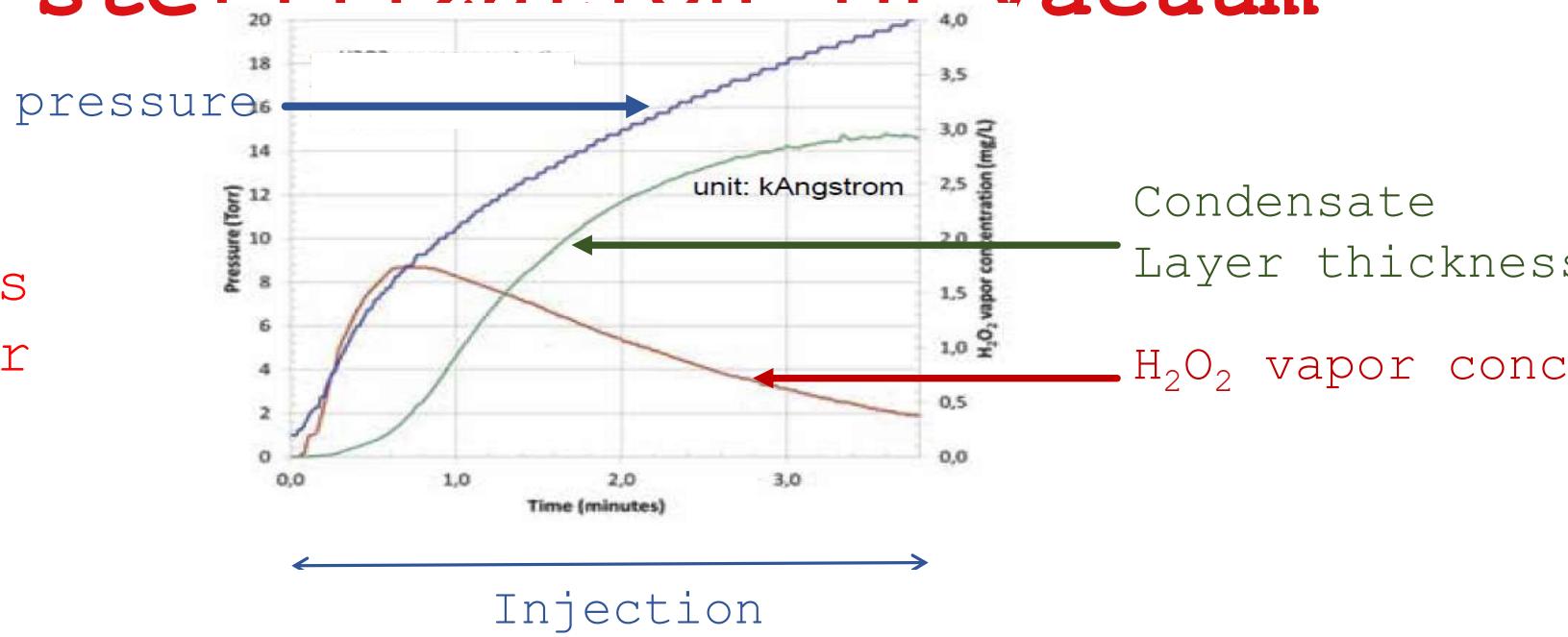
$e \uparrow$ ($RH_i \uparrow$ H_2O_2 ppm \downarrow Eff \downarrow)
Dilution



Adsorption / condensation
 $e_{es} = F(RH_i, H_2O_2$
 ppm)

Comparison with H₂O₂ sterilization in vacuum

Timescale is
much shorter



(pulses of H₂O₂ to reach 19 Torr – specific to this process)

The first dual-sterilant low-temperature sterilization system; from Dufresne and Richards, Canad. J. of Infection Control (2016)

Concluding remarks

- Vacuum to withdraw non-condensable gases
- Fast and efficient penetration in instruments of the sterilizing agent in vapor state
- Condensation/adsorption of water needed for various reasons for each sterilization modalities
- ... But:
 - Condensation is a complex phenomenon, not to be reduced to p, V, T ...
 - Growing complexity of medical devices

THANK YOU !